

Edmore Public School 706 Main St, Edmore, ND 58330

Chemistry Lesson Plans for February 6 - 10, 2023 1st Hour, 8:40 – 9:32 AM

	Monday (Feb 6)	Tuesday (Feb 7)	Wednesday (Feb 8)	Thursday (Feb 9)	Friday (Feb 10)
Performance	HS-PS1-7	HS-PS1-7	HS-PS1-7	HS-PS1-7	
Standards	Use mathematical	Use mathematical	Use mathematical	Use mathematical	
	representations to support	representations to support	representations to support	representations to support	
	the claim that atoms, and	the claim that atoms, and	the claim that atoms, and	the claim that atoms, and	
	therefore mass, are	therefore mass, are	therefore mass, are	therefore mass, are	
	conserved during a chemical	conserved during a chemical	conserved during a chemical	conserved during a chemical	
	reaction.	reaction.	reaction.	reaction.	
Topic	Excess Reactant - Simulation	Excess Reactant - Explanation	Excess reactant - Extension	Percent Yield	
Objectives	• Identify and calculate the mass and moles of the excess reactant in a chemical reaction.	• Identify and calculate the mass and moles of the excess reactant in a chemical reaction.	• Identify and calculate the mass and moles of the excess reactant in a chemical reaction.	• Identify and calculate the mass and moles of the excess reactant in a chemical reaction.	
Bellringer	(3 min) molecular formula	(3 min) empirical formula	(3 min) excess reactant	(3 min) percentage yield	
Procedure/ Instructional Delivery	excess reactant simulation activity	 direct instruction: discussion on excess reactant and solving sample problems Guided practice: solving for excess reactant 	Independent practice: students will work independently on excess reactant worksheet	Case analysis: importance of percentage yield reading: main concepts of percentage yield	
Assessment	worksheet	worksheet	worksheet	worksheet	
Remarks					No School

Prepared by:

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